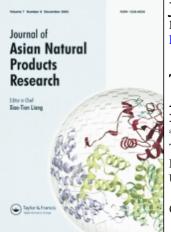
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# Two new dimeric secoiridoid glycosides from the fruits of *Ligustrum*

lucidum

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### **ORIGINAL ARTICLE**

### Two new dimeric secoiridoid glycosides from the fruits of *Ligustrum lucidum*

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Two new secoiridoid glycosides, ligusides A and B (1 and 2), as well as seven known compounds (3-9), were isolated from the fruits of *Ligustrum lucidum*. Their structures were elucidated on the basis of spectroscopic and chemical analysis.

Keywords: Ligustrum lucidum; Oleaceae; secoiridoid glycoside; ligusides A and B

#### 1. Introduction

Ligustrum lucidum Ait. (Oleaceae), a wellknown medicinal plant, is distributed widely in China and Southeast Asia. The fruits of this plant were used comprehensively as a tonic agent for liver and kidney protection in the traditional Chinese medicine. The aqueous extract of the fruits showed significant antioxidant [1], antiviral [2], anticancer [3], and antisenile [4] activities. Previous phytochemical research of the fruits afforded plenty of oleoside-type secoiridoid glycosides in the water-soluble fraction [5,6]. In our previous work [7], 15 triterpenes were isolated from the ethanol extract of the fruits of L. lucidum. In continuation of our chemical investigation on this plant, two new secoiridoid glycosides, ligusides A(1)and B (2), together with seven known compounds, 8(E)-nuezhenide (3) [8], Gl3 (4) [9],  $1'''-O-\beta$ -D-glucosylformoside (5) [10], oleuropeinic acid (6) [11], oleoside dimethyl ester (7) [12], 8(Z)-nuezhenide A (8) [13], and oleonuezhenide (9) [14] (Figure 1), were isolated. In this paper, we report the isolation and structural determination of the two new secoiridoid glycosides.

#### 2. Results and discussion

Compound 1 was obtained as an amorphous powder. Its molecular formula was determined as C48H64O27 by HR-ESI-MS at m/z 1095.3525 [M + Na]<sup>+</sup>. The UV spectrum showed absorption maxima at 236 nm. The IR spectrum of 1 showed the presence of hydroxyl  $(3413 \text{ cm}^{-1})$ , carbonyl  $(1701 \text{ cm}^{-1})$ , and olefinic bond  $(1631 \text{ cm}^{-1})$ . The presence of D-glucopyranosyl was confirmed by acid hydrolysis of 1 with 1 M HCl, followed by the GC analysis comparing with authentic samples. The <sup>1</sup>H NMR spectrum of **1** exhibited signals for two oleoside 11methyl ester moieties [H-3 at  $\delta_{\rm H}$  7.56 and 7.52 (each 1H, s), H-8 at  $\delta_{\rm H}$  6.18 and 6.14 (each 1H, q, J = 7.0 Hz), H-1 at  $\delta_{\rm H}$  6.02 and 5.96 (each 1H, s), H-1' at  $\delta_{\rm H}$  4.82

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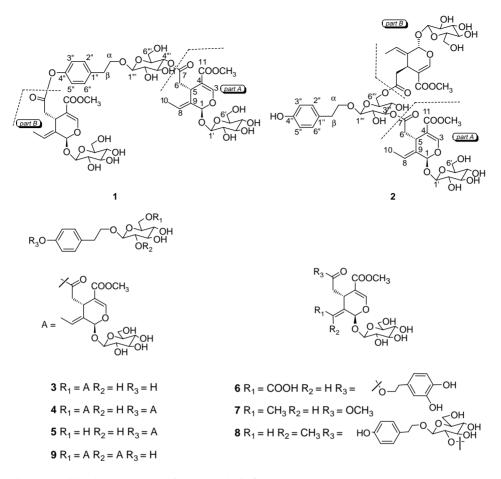


Figure 1. Chemical structures of compounds 1-9.

(1H, d, J = 7.9 Hz) and 4.80 (1H, d,J = 8.1 Hz), OMe at  $\delta_{\text{H}}$  3.73 and 3.71 (each 3H, s), H<sub>3</sub>-10 at  $\delta_{\rm H}$  1.76 (6H, d,  $J = 7.0 \,\mathrm{Hz}$ ] [9], a *p*-hydroxyphenethoxyl moiety [ $\delta_{\rm H}$  7.29 (2H, d, J = 8.5 Hz), 6.99 (2H, d, J = 8.5 Hz), 4.08 (1H, m), 3.77(1H, m), 2.93 (2H, m)], as well as an additional anomeric proton [ $\delta_{\rm H}$  4.33 (1H, d,  $J = 7.7 \,\text{Hz}, \,\text{H-1}^{\prime\prime\prime}$ ]. The coupling constant of the anomeric proton indicated the presence of  $\beta$ -configuration of the D-glucose unit. In accordance with the above findings, the  ${}^{13}C$  NMR and DEPT spectra of 1 showed carbon signals corresponding to two secoiridoid glucoside units, together with the resonances of a *p*-hydroxyphenethoxyl and an additional glucose moiety. With the aid of the 2D NMR (<sup>1</sup>H-<sup>1</sup>H COSY, HSOC, and HMBC) spectra, all the <sup>1</sup>H and <sup>13</sup>C NMR signals of 1 were assigned and are shown in Table 1. The <sup>1</sup>H and <sup>13</sup>C NMR spectral data of 1 were closely similar to those of Gl3 (4) [9], except for the signals arising from the additional glucose unit, which indicated that an oleoside 11-methyl ester unit could be attached to different positions of the glucose. The HMBC correlation between H-4<sup>'''</sup> ( $\delta_{\rm H}$  4.67) and C-7 ( $\delta_{\rm C}$  172.4, in part A) suggested that the oleoside 11-methyl ester unit (part A) was esterified with the hydroxyl group at C-4<sup>III</sup> of an additional glucose moiety. In addition, the HMBC correlation between H-1<sup>///</sup> and  $C-\alpha$  suggested that the additional glucose

		1				2		
	8	δ <sub>H</sub>	Ş,	$\delta_{\rm C}$	~	bн	\$	δc
No.	Part A	Part B	Part A	Part B	Part A	Part B	Part A	Part B
-	5.96 (s)	6.02 (s)	95.5	95.4	5.92 (s)	5.96 (s)	95.1	95.6
3	7.52 (s)	7.56 (s)	155.3	155.4	7.52 (s)	7.53 (s)	155.2	155.3
4	1	1	109.3	109.3	1	I	109.3	109.4
5	4.02  (dd, J = 4.0, 9.8)	4.11	31.6	31.8	4.01	4.01	31.6	31.8
9	2.57  (dd,  J = 9.8, 14.9)	2.73 (dd, $J = 9.2$ , 14.4)	40.7	41.0	2.74	2.74	40.7	41.3
	$2.85 (\mathrm{dd}, J = 4.0, 14.9)$	2.96				2.50		
L	1	1	172.4	171.7	1	1	172.8	172.9
8	6.18 (q, $J = 7.0$ )	6.14  (q, J = 7.0)	125.4	125.1	6.09 (q, $J = 7.1$ )	6.12 (q, $J = 7.1$ )	124.9	125.1
6	1	1	130.4	130.6	I	1	130.5	130.5
10	1.76 (d, $J = 7.0$ )	1.76 (d, $J = 7.0$ )	13.8	13.8	$1.72 (\mathrm{dd}, J = 1.4, 7.1)$	$1.77 (\mathrm{dd}, J = 1.5, 7.1)$	13.7	13.8
11	I	I	168.7	168.7	I	I	168.9	168.9
$OCH_3$	3.71 (s)	3.73 (s)	52.0	52.0	3.69 (s)	3.73 (s)	52.0	52.0
1'	4.82 (d, $J = 7.9$ )	4.80 (d, $J = 8.1$ )	101.2	101.0	4.81 (d, $J = 7.8$ )	4.81 (d, $J = 7.8$ )	100.8	100.9
2	3.33	3.33	74.6	74.7	3.31	3.31	74.7	74.7
3/	3.33	3.33	78.3	78.3	3.41	3.41	<i><b>77.9</b></i>	<i>0.17</i>
4	3.33	3.33	71.2	71.4	3.27	3.27	71.5	71.6
5'	3.40	3.40	<i><b>77.9</b></i>	77.9	3.33	3.33	78.4	78.4
6'	3.86 (t, J = 13.2)	3.86 (t, J = 13.2)	62.4	62.5	$3.88 (\mathrm{dd}, J = 2.1, 12.1)$	$3.90 (\mathrm{dd}, J = 2.0, 12.2)$	62.8	62.8
	$3.69 (\mathrm{d}, J = 11.4)$	3.69 (d, J = 11.4)			3.66	3.66		
α	3.77			71.6	3.96			72.3
	4.08				3.72			
ප	2.93			36.5	2.84			36.4
$1^{\prime\prime}$	1			137.9	I			130.6
2", 6" 2", 5"	7.29 (d, $J = 8.5$ )			131.0	7.04  (d, J = 8.5)			130.9
с, с ″/	$(c.\delta = t, u) \ ee{0}$			150.5	$(c.\delta = t, \mu) \delta 0.0$			156.8
1,1	= 4.33 (d. $J = 7.7$ )			104.2	= 4.38 (d. $J = 7.8$ )			104.3
2'''	3.26			75.1	3.32			73.1

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR spectral data of compounds 1 and 2 (CD<sub>3</sub>OD, J in Hz)<sup>a,b</sup>.

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687

1 ô <sub>H</sub> Part B		δ <sub>H</sub> Part A 4.90 3.43 3.53 3.53 4.23 (dd, J = 5.4, 11.9)	2 Part B	δ <sub>C</sub> Part A Part B 78.9 69.8 64.6	δ <sub>C</sub> A Part B 78.9 69.8 64.6
	1 ô <sub>H</sub> Part B	$\frac{1}{\delta_{H}} \frac{\delta_{C}}{Part B} \frac{\delta_{C}}{Part A P}$	$\begin{array}{c c} \mathbf{\hat{a}}_{\mathrm{H}} & \mathbf{\hat{b}}_{\mathrm{C}} \\ \hline & & & & & & \\ \hline & & & & & & \\ & & & &$	$1$ $\delta_{\rm C}$ $\delta_{\rm C}$ $\delta_{\rm H}$ $\delta_{\rm H}$ $\delta_{\rm C}$ $\delta_{\rm H}$ Part BPart APart A75.872.93.4372.93.5362.64.23 (dd, $J = 5.4, 11.9)$ 4.31 (dd, $J = 2.1, 11.8)$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

moiety was substituted by the *p*-hydroxyphenethoxyl unit at  $C-1^{\prime\prime\prime}$ . Therefore, the structure of **1** was established and named as liguside A.

Compound 2 was obtained as an amorphous powder. Its molecular formula was determined to be C48H64O27 by HR-ESI-MS at m/z 1071.3547 [M - H]<sup>-</sup>. Acid hydrolysis of 2 also yielded D-glucose, which was identified by the GC analysis. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **2** showed signals corresponding to two oleoside 11methyl ester units, together with the resonances of a *p*-hydroxyphenethoxyl moiety and an additional glucose moiety as observed in 1. The combined analysis of 1D and 2D NMR spectra led to the assignment of all the <sup>1</sup>H and <sup>13</sup>C NMR spectral signals of 2 (Table 1). The <sup>1</sup>H and  $^{13}$ C NMR data of **2** were closely similar to those of oleonuezhenide (9) [14]. The main differences were the signals assigned to the additional glucose unit. In the HMBC spectrum of 2, the correlations between H-3''' ( $\delta_{\rm H}$  4.90) and C-7 ( $\delta_{\rm C}$  172.8, in part A), as well as between H-6<sup>*III*</sup> ( $\delta_{\rm H}$  4.23 and 4.31) and C-7 ( $\delta_{\rm C}$  172.9, in part B), suggested that the two oleoside 11-methyl ester units were substituted at the C-3" and C-6" positions of the glucose, respectively. Thus, the structure of 2 was assigned and named as liguside B.

#### 3. Experimental

#### 3.1 General experimental procedures

Melting points were recorded on an XT-4 micromelting point apparatus without correction. Optical rotation values were measured on a JASCO P-1020 digital polarimeter with a 0.01 dm length cell. IR spectra (KBr) were recorded on a JASCO FT/IR-480 plus Fourier transform infrared spectrometer. UV spectra were obtained on a JASCO V-550 UV/vis spectrophotometer with a 0.1 dm length cell. <sup>1</sup>H (400 or 500 MHz), <sup>13</sup>C (100 or 125 MHz), and 2D NMR spectra were recorded on Bruker AV-400 or AV-500 spectrometers. HR-ESI-MS

Table 1 – continued

spectra were measured using an Agilent 6210 LC/MSD TOF mass spectrometer. Preparative HPLC was carried out on a Varian instrument equipped with UV detectors (Varian, Palo Alto, CA, USA) and a reversed-phase  $C_{18}$  column (5  $\mu$ m,  $20 \text{ mm} \times 250 \text{ mm}$ ; Cosmosil, Houston, TX, USA). Column chromatographies were carried out using silica gel (200-400 mesh; Qingdao Haiyang Chemical Group Corporation, Qingdao, China) and Sephadex LH-20 (Pharmacia Biotech AB, Uppsala, Sweden). TLC analysis was performed on precoated silica gel GF<sub>254</sub> plates (Yantai Chemical Industrial Institute, Yantai, China).

#### 3.2 Plant material

The fruits of *L. lucidum* were collected in Nanjing City, Jiangsu Province of China, in October 2006, and were authenticated by Prof. Min-Jian Qin (Department of Pharmacognosy, China Pharmaceutical University, Nanjing, China). A voucher specimen (No. 20060517) was deposited in the herbarium of the China Pharmaceutical University, Nanjing.

#### 3.3 Extraction and isolation

Dried and roughly powdered fruits of L. lucidum (40 kg) were refluxed with 70% EtOH  $(3 \times 40 \text{ liters}, 1 \text{ h each})$ . The EtOH extract was concentrated in vacuo to yield a crude extract (3.2 kg), which was redissolved in water and partitioned successively by petroleum ether, EtOAc, and n-BuOH, respectively. The n-BuOH layer was concentrated under vacuum to afford a residue (1.2 kg), which was subjected to a macroporous resin D101 (3 kg) column, and eluted with EtOH $-H_2O$  (0:100, 15:85, 35:65, 55:45, 95:5). The 35% EtOH extract (200 g) was separated by silica gel column chromatography (200-300 mesh, 1.5 kg) using gradient mixtures of CHCl3-MeOH-H<sub>2</sub>O (90:10:0.1-50:50:5) as eluants to afford 11 fractions (Fr. 1-11).

Fr. 7 (10g) was subjected to an ODS column using gradient mixtures of MeOH-H<sub>2</sub>O (28:72, 30:70, 35:65, 40:60, 50:50, and 70:30, each 1000 ml) as eluants to yield 10 subfractions. Compound 3(2g)was obtained upon recrystallization in MeOH-H<sub>2</sub>O (1:1) from subfraction 7-3 (4 g). Subfraction 7-5 (1 g) was subjected to a preparative HPLC (MeOH-H<sub>2</sub>O, 35:65) to give 1 (10 mg), 4 (40 mg), 5 (300 mg), 6 (12 mg), 7 (5 mg), and 9 (10 mg), respectively. Subfraction 7-8 (2.3 g) was purified by a preparative HPLC (MeOH-H<sub>2</sub>O, 45:55) to afford 2 (11 mg). Fraction 8 (12 g) was also subjected to a preparative HPLC (MeOH $-H_2O$ , 30:70) to yield 8 (9 mg).

#### 3.3.1 Compound 1

An amorphous powder, mp 130–131°C;  $[\alpha]_{D}^{20}$ –169.5 (c = 0.2, MeOH); UV (MeOH)  $\lambda_{max}$ : 236 nm; IR (KBr)  $\nu_{max}$ : 3413, 1701, 1631, 1516, 1075, 1019 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; HR-ESI-MS: m/z 1095.3525 [M + Na]<sup>+</sup> (calcd for C<sub>48</sub>H<sub>64</sub>O<sub>27</sub>Na, 1095.3527).

#### 3.3.2 Compound 2

An amorphous powder, mp 128–130°C;  $[\alpha]_{D}^{20}$ –100.5 (c = 0.2, MeOH); UV (MeOH)  $\lambda_{max}$ : 231 nm; IR (KBr)  $\nu_{max}$ : 3424, 2853, 2922, 1722, 1633, 1519, 1444, 1310, 1182, 1079, 1043 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; HR-ESI-MS: m/z 1071.3547 [M – H]<sup>-</sup> (calcd for C<sub>48</sub>H<sub>63</sub>O<sub>27</sub>, 1071.3562).

# 3.4 Acid hydrolysis and GC analysis of 1 and 2

Compounds 1 and 2 (each 1 mg) were refluxed with 1 M HCl in 60% EtOH (1 ml) at 95°C for 2 h, respectively. Each reaction mixture was evaporated with a stream of  $N_2$ to yield a residue, which was then dissolved in H<sub>2</sub>O and extracted with CHCl<sub>3</sub>. The aqueous layer was concentrated and dried by N<sub>2</sub>, and treated with pyridine (1 ml) and L-cysteine methyl ester hydrochloride (3 mg), followed by heating at 60°C for 2 h, and then concentrated to dryness with N<sub>2</sub>. The residue was added to N-(trimethylsilyl) imidazole (0.2 ml) and kept at 60°C for 1 h. Subsequently, the solution was diluted with H<sub>2</sub>O (1 ml) and extracted with cyclohexane (2 ml). The supernatant was subjected to GC analysis, [Shimadzu GCMS-QP2010 plus gas chromatograph/mass spectrometer; column: HP-1701  $(0.25 \text{ mm} \times 30 \text{ m})$ , detector: FID, column temperature: 200-250°C (5°C/min), detector temperature: 280°C, injector temperature: 250°C, carrier gas: N<sub>2</sub>]. The standard D-glucose and L-glucose were subjected to the same reaction and GC analysis under the above conditions [retention times (min): 32.289 (D-glucose), 34.874 (Lglucose)]. For 1-2, D-glucose was detected.

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